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S Supporting Information

[AB](#page-9-0)STRACT: [In a fresh a](#page-9-0)pproach to the synthesis of Nvinylazoles, a ligand-free palladium catalytic system was found to promote the Csp^2-N bond-forming reaction utilizing Ntosylhydrazones and N-H azoles. This process shows functional group tolerance; di-, tri-, and tetrasubstituted Nvinylazoles were obtained in high yields. Under the optimized conditions, the reaction proceeds with high stereoselectivity depending on the nature of the coupling partners.

ENTRODUCTION

N-Vinylazoles are important classes of building blocks in organic synthesis and are also key structural motifs in medicinal chemistry.¹ They have been found to display antifungal activity. 2 N-Vinylazoles have been shown to serve as monomers for the s[yn](#page-9-0)thesis of poly(N-vinylazoles).³ These latter have been u[til](#page-9-0)ized as semiconductors and photosensitive materials. There are several methods for the prepara[ti](#page-9-0)on of N-vinylazoles. The most conventional route is the condensation of N-H azoles with carbonyl compounds in the presence of a water scavenger and a Brønsted or Lewis acid (Scheme 1a).⁴ The relatively harsh reaction conditions generally required for this transformation cause low functional group tole[ra](#page-1-0)n[ce](#page-9-0) and prompted the emergence of alternative methods.⁵ Recently, with the development of modern organometallic chemistry, transitionmetal-catalyzed coupling reactions o[ff](#page-9-0)er a more reliable approach to the preparation of N-vinylazoles by coupling N-H azoles with vinyl halides or vinyl triflates (Scheme 1b). 1,6 The major limitation of this method is that multiple steps are generally required for the preparation of the vinyl hal[id](#page-1-0)e. [A](#page-9-0) more attractive approach is the amination of alkynes (Scheme 1c).⁷ Intermolecular addition of amines to alkynes has been well-studied.⁸ However, the addition of N-heterocycles onto alk[yn](#page-9-0)es remains elusive, and the regioselectivity issue is always a [d](#page-1-0)aunting tas[k](#page-9-0).^{7a,9} Therefore, it would be desirable to develop a new type of coupling reaction to form N-vinylazoles that may circumvent t[hese](#page-9-0) drawbacks.¹⁰

Over the past years, N-tosylhydrazones have attracted extensive attention because [of](#page-9-0) their various useful applications in organic synthesis. In particular, they are valuable and readily available reagents in C−C,¹¹ C−S,¹² C−O,¹³ and C−B¹⁴ bondforming reactions through metal-catalyzed and metal-free processes. In this area, w[e r](#page-9-0)eport[ed](#page-9-0) the C[u-](#page-9-0)catalyzed [Cs](#page-9-0)p³ $-$ N

bond-forming reaction between N-tosylhydrazones and aliphatic amines, giving rise to the reductive coupling products (Scheme 2a).¹⁵ Herein, we further report the oxidative Pdcatalyzed cross-coupling of N-tosylhydrazones and N-H azoles, which co[ns](#page-1-0)tit[ute](#page-9-0)s a highly efficient and practical approach for Csp²−N bond formation (Scheme 2b).

In the course of our interest of sulfonylhydrazones as versatile coupling partners,¹⁶ very [r](#page-1-0)ecently, we developed a novel Pd-catalyzed three-component reaction (MCR) between N-tosylhydrazones, dihaloa[ren](#page-9-0)es, and amines (e.g., anilines, aliphatic amines), producing nitrogen-containing 1,1′-diarylethylenes of biological interest through a faster $C=C$ bond formation and an efficient intermolecular C−N crosscoupling.¹⁷ To expand the scope of this catalytic MCR to a wider variety of new coupling partners (e.g., N-H azoles), we decided [to](#page-10-0) study the employ of indoles 2 as nucleophilic components in the cross coupling process of N-tosylhydrazone 1a and 1-chloro-4-iodobenzene 3a. However, to our surprise, under optimized conditions, the 1,1′-diarylethylene containing an indole unit 5 was never detected and, instead, N-vinylindole 4a was isolated in a good 70% yield (Scheme 3).

On the basis of this unexpected result, we have explored this new type coupling reaction that allows an expe[di](#page-1-0)tious and easy access to N-vinylindole derivatives. To the best of our knowledge, this is the first report describing the palladiumcatalyzed Csp²-N bond formation through the oxidative coupling of N-H azoles and N-tosylhydrazones (Scheme 2b). Mechanistically, this Csp^2-N bond-forming reaction is fundamentally different from those classically reported in [t](#page-1-0)he literature.⁵ Moreover, the use of N-tosylhydrazones represents a

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Scheme 1. Synthetic Approaches Toward N-Vinylindoles

Scheme 2. Transition-Metal-Catalyzed C−N Bond-Forming Reactions of N-Tosylhydrazones with Amines

very convenient methodology for the unconventional modification of carbonyl compounds.

■ RESULTS AND DISCUSSION

We began the exploration of this new transformation with Ntosylhydrazone 1a and indole 2a as a model substrate (Table 1; see the Supporting Information for the complete study). Since the final product 4a of the MCR depicted in Scheme 3 did n[ot](#page-2-0) incorpo[rate the aryl unit of](#page-9-0) 3a, we carried out the coupling, without the addition of halogenated derivative 3a (Table 1, entry 1). Under these conditions, no trace of 4a was detected. In this case, thermolysis of hydrazone 1a provides t[he](#page-2-0) formation of two concomitant byproducts: (i) the Bamford− Stevens¹⁸ alkene $(1,3,5$ -trimethoxy-2-vinylbenzene), resulting from the evolution of the diazo intermediate, and (ii) the reducti[ve](#page-10-0) etherification product (2-(1-(tert-butoxy)ethyl)-1,3,5 trimethoxybenzene)¹³ derived from the reaction between the carbene complex and base (NaOtBu). This result suggests that 3a plays the role of [th](#page-9-0)e oxidant for this coupling. For the next experiments, we found that simply using iodobenzene as the oxidant serves as an effective alternative to 3a. Next, the reaction without ligand Xphos was performed (entry 2).

Gratifyingly, the desired N-vinylindole 4a was obtained as the sole reaction product in a very promising 75% isolated yield. Other combinations of palladium source, base, and solvents were then examined. The base of choice for this transformation was NaOtBu (see, entries 2−4). On the basis of the results obtained in entries 2 and 6, commercially available $Pd_2(dba)$ ₃· CHCl₃ was fixed as the source of palladium. Screening of the solvent source demonstrates that fluorobenzene (PhF) and cyclopentylmethyl ether (CPME) give the best results (entries 6 and 9). Under optimal conditions, clean and full conversion of the starting material was achieved to give 4a in nearly quantitative yield upon isolation (Table 1, entry 9). Notably, a low conversion was observed when iodobenzene was changed to bromobenzene, and no reaction occu[rre](#page-2-0)d in the presence of O2 or 1,4-benzoquinone (1,4-BQ) (entries 11−12). It should be noted that the coupling of hydrazone 1a with indole 2a is not limited to a small scale (0.75 mmol) as it could be conveniently performed on a 2.5 g scale for 1a (6.6 mmol), giving rise to 4a in 90% yield.

We next explored the scope of this useful cross-coupling of N-tosylhydrazones and indoles (Table 2). The reaction is general with various hydrazones derived from acetophenones and 4,5,6,7-substituted indoles (compo[un](#page-3-0)ds 4b−4m). The reaction displays no dependence upon the electronic nature and position of the substituent on the aromatic ring of the indole or hydrazone moiety. Electron-rich and electron-poor indoles or hydrazones all reacted completely and effectively within 1 h (compounds 4b−4i). It is noteworthy that functional groups, such as amine, nitrile, and alkyne, are welltolerated (compounds 4k−4m).

Scheme 3. Pd(II)-Catalyzed Csp²−N Bond Formation through the Oxidative Coupling of Indole 2a and N-Tosylhydrazone 1a

a
Reaction conditions: N-tosylhydrazone 1a (0.75 mmol), indole 2a (0.5 mmol), additive (0.6 mmol), [Pd] (2 mol %), base (1.4 mmol), solvent (4 mL) at reflux for 2 h. ^bYield of isolated product 4a. Coupling was performed in the presence of Xphos ligand (4 mol %). ^dIn this case, coupling between hydrazone 1a and iodobenzene gives the corresponding olefin, mainly (1,3,5-trimethoxy-2-(1-phenylvinyl)benzene).

To further expand the scope of this reaction, we studied the coupling with various hydrazone partners (compounds 4n− 4v). As depicted in Table 2, the coupling of N-tosylhydrazones derived from aliphatic ketones, aldehydes, chromanones, and tetralones with different indoles led to the expected Nvinylindole products in excellent yields. The reaction was also extended to hindered tosylhydrazones. Substrates containing a secondary carbon atom α to the hydrazone function were successfully coupled with 5-halo-indole, to provide tetrasubstituted vinylindoles 4q and 4r having a cycloalkylidene unit in good yields.

In this study, we also examined the stereoselectivity issue. Interestingly, the coupling reaction of a hydrazone derived from propiophenone, which features a methoxy group in the orthoposition, afforded mainly the Z-olefin $4m$ (E/Z 10/90). More interestingly, coupling between indoles and hydrazones derived from 1,2-diphenylethanone compounds afforded exclusively the Z-vinylindoles 4s and 4t, whereas the reaction with the hydrazone derived from 3-pentanone provided a 45/55 mixture of the E/Z isomers (compound 4o).

To highlight the power of this vinylation of indoles, the Pdcatalyzed Csp^2-N bond formation reaction was applied to other types of substrates (Table 3). 2- or 3-Substituted indoles showed excellent reactivity (compounds 4w−4ae). Gratifyingly, sterically hindered N-tosylhydr[az](#page-3-0)ones featuring ortho/ortho′ substituents on the aromatic ring could be installed efficiently (compounds 4w, 4x, 4z, and 4ab). The use of 2,3-disubstituted indole also delivered the corresponding product in a good 88% yield (compound 4ae). It should be noted that the chemoselectivity of this reaction must be underlined since tryptamine was selectively coupled with hydrazone, giving rise to compound 4aa in a good 78% yield.

Encouraged by the results obtained with substituted indoles, we proceeded to apply this catalytic system to other N-H azoles. To our delight, it was found that the reaction is applicable to carbazoles (compounds 4af−4ah), including 3,6 dibromocarbazole, which can be subjected to a further functionalization (C−Br bonds). The resulting compound 4ah may be used for the synthesis of some multiaryl compounds by stepwise cross-coupling reactions.¹⁰

Also, coupling worked well for other N-H azoles, such as 1,2,3,4-tetrahydrocarbazole, pyrrole, 1,5,6,7-te[tra](#page-9-0)hydro-4Hindol-4-one, benzimidazole, and imidazole. Thus, the corresponding N-vinylazoles (compounds 4ai−4an) were obtained in yields ranging from 35% to 80%.

A plausible mechanism for this Pd-catalyzed oxidative coupling is depicted in Scheme 4. Iodobenzene acts as a viable oxidant for this transformation, and its dehalogenation leads to the formation of unreactive be[nz](#page-4-0)ene that becomes part of the solvent.¹⁹ The reaction is initiated by the oxidation of $Pd(0)$ to Pd(II) species by iodobenzene, which reacts with the in situ generat[ed](#page-10-0) diazo substrate I to give Pd–carbene complex III.²⁰ Ligand exchange between complex III and indole leads to the formation of species IV, which undergoes a migratory inserti[on](#page-10-0) of the indole unit to furnish the alkyl palladium complex V. Further β -hydride elimination provides the cross-coupling product 4 and species VI, which regenerates the $Pd(0)$ catalyst after reductive elimination.

To obtain additional information on the reaction mechanism, the coupling between d_2 -deuterated 6^{21} and indole 2a has been performed under our optimized catalytic conditions using 3,4,5 trimethoxyiodobenzene 7 as the oxi[da](#page-10-0)nt instead of iodobenzene. According to the mechanism depicted in Scheme 4, we can expect the formation, through a reductive elimination step, of trimethoxydeuterobenzene 8. As shown in Scheme [5](#page-4-0), as expected, the reaction product 4ao isolated from this experiment contains a significant amount of deuterium [at](#page-4-0) the vinyl carbon $β$ to indole. In addition, we isolated compound 8 in which the carbon−I bond of 7 was replaced by a C− deuterium bond. This result is an agreement with the formation of 4 by β -hydrogen elimination and the role played by aryl iodide 7 as the oxidant.

The Z-stereoselectivity observed for compounds 4r and 4s was rationalized by performing a computational chemistry study.²² Specifically, we were interested in the $syn-\beta$ -hydrogen elimination step, which would be involved in the control of the

 a Reaction conditions: N-tosylhydrazone 1 (0.75 mmol), indole 2 (0.5) mmol), PhI (0.6 mmol), $Pd_2(dba)_3$ ·CHCl₃ (2 mol %), NaOtBu (1.4 mmol), CPME (4 mL) at reflux for 1 h. $\frac{b}{b}$ Yield of isolated product.

coupling, with unprotected S-hydroxyindele does not proceed Coupling with unprotected 5-hydroxyindole does not proceed. d Compound 4m was obtained as a 10:90 mixture of E/Z isomers. Compound 4o was obtained as a 45:55 mixture of E/Z isomers. Compounds 4s and 4t were obtained as single Z isomers. E/Z ratio was determined by ¹HNMR.

stereochemistry of the double bond. Starting from the alkyl palladium complex A (Scheme 6), we compute for the β - Table 3. Scope of the Pd-Catalyzed Cross-Coupling of Hydrazones 1 and Indole Derivatives $2^{a,b}$

a
Reaction conditions: N-tosylhydrazone 1 (0.75 mmol), indole 2 (0.5 mmol), PhI (0.6 mmol), $Pd_2(dba)_3$ ·CHCl₃ (2 mol %), NaOtBu (1.4 mmol), CPME (4 mL) at reflux for 1 h. $\frac{b}{b}$ Yield of isolated product.

hydrogen elimination step both the transition states TS-Z and TS-E that lead to the Z and E double bonds, respectively (isomers B-Z and B-E). The examination of the molecular models of transition states shows similar arrangements of the phenyl and indole substituents. In both cases, the indole moeity (in red in Scheme 6) is almost orthogonal relative to the plane formed by the incipient double bond, with its six-membered

ring anti to the metal. This conformation minimizes the steric interaction for indole and induces an almost coplanar position of the phenyl group (in blue in Scheme 6) and the incipient double bond. In this situation, the steric hindrance of the phenyl group (in blue) close to the in[cip](#page-5-0)ient double bond exceeds that of the indole (in red). The phenyl substituent located on the alkyl chain (in green in Scheme 6) has, therefore, lower steric interaction when located trans to this phenyl group (in blue). Consequently, the transiti[on](#page-5-0) state leading to the Z isomer (TS-Z) is 2.5 kcal·mol⁻¹ lower in energy than TS-E, explaining that the formation of the Z isomer is clearly favored.

■ CONCLUSION

In summary, we have described a new procedure for the preparation of N-vinylazole compounds by free-ligand Pdcatalyzed cross-coupling between N-tosylhydrazones and various azole reagents. This reaction, which involves an unprecedented indole migratory insertion of a palladium carbene, was used to obtain a variety of N-vinylazoles, including trisubstituted compounds, in a stereoselective manner. Unlike established methods for N-vinylazole formation, the current methodology requires no additional organometallic reagent. Moreover, the N-tosylhydrazones used are readily available from the corresponding ketones or aldehydes and are easy to handle. All of these features make this method a useful extension of palladium-catalyzed coupling reactions for Nvinylazole synthesis.

EXPERIMENTAL SECTION

General Methods. Solvent peaks were used as reference values, with $CDCl₃$ at 7.26 ppm for ¹H NMR and 77.16 ppm for ¹³C NMR. Chemical shifts δ are given in parts per million, and the following abbreviations are used: singlet (s), doublet (d), doublet of doublet (dd), triplet (t), multiplet (m), and broad singlet (bs). Reaction courses and product mixtures were routinely monitored by TLC on silica gel, and compounds were visualized with phosphomolybdic acid/ Δ, anisaldehyde/Δ, or vanillin/Δ. Flash chromatography was performed using silica gel 60 (40−63 mm, 230−400 mesh) at medium pressure (200 mbar). Fluorobenzene was used as received; dioxane, dichloromethane, cyclohexane, and tetrahydrofuran were dried using the procedures described in Purification of Laboratory Chemicals.²³ Organic extracts were, in general, dried over MgSO₄ or $Na₂SO₄$. High-resolution mass spectra were recorded with the aid of a MicrOT[OF-](#page-10-0)Q II. All products reported showed ¹H and ¹³C NMR spectra in agreement with the assigned structures.

General Procedure for Preparation of Hydrazones. 24 To a rapidly stirred suspension of p-toluenesulphonohydrazide (5 mmol) in dry methanol (10 mL) at 60 °C, the ketone (5 mmol) w[as](#page-10-0) added dropwise. Within 5−60 min, the N-tosylhydrazone began to precipitate. The mixture was cooled to 0 °C, and the product was collected on a Bü chner funnel, washed with petroleum ether, and then dried in vacuo to afford the pure product. The reaction provides the Ntosylhydrazone derivatives in about 88−99% yields.

Typical Pd-Catalyzed Oxidative Cross-Coupling of Hydrazones and Indole Derivatives. A 10 mL round-bottom flask with a condenser under an argon atmosphere was charged with Ntosylhydrazone (1.5 equiv), iodobenzene (1.2 equiv), $Pd_2(dba)$ ₃· CHCl₃ (2 mol %), NaOtBu (2.8 equiv), and indole (1 equiv). A 4 mL portion of CPME was then added via syringe at room temperature. The flask was put into a preheated oil bath and stirred at reflux for 1 h. The crude reaction mixture was allowed to cool to room temperature. EtOAc was added to the mixture, which was filtered through Celite. The solvents were evaporated under reduced pressure, and the crude residue was purified by flash chromatography on silica gel.

1-(1-(2,4,6-Trimethoxyphenyl)vinyl)-1H-indole 4a. Flash chromatography on silica gel (EtOAc/cyclohexane 2/98) afforded 150 mg of 4a (0.49 mmol, yield 97%); white solid, mp: 134−136 °C; TLC: R_f = 0.58 (EtOAc/cyclohexane, 20/80, SiO₂); IR (neat) 1641, 1604, 1582, 1453, 1414; ¹H NMR (300 MHz, CD_3COCD_3) δ (ppm) 7.52 (m, 1H), 7.36 (m, 1H), 7.17 (d, J = 3.3 Hz, 1H), 7.02 (m, 2H), 6.44 (d, J = 3.3 Hz, 1H), 6.29 (s, 2H), 5.56 (s, 1H), 5.08 (s, 1H), 3.85 (s, 3H), 3.66 (s, 6H); ¹³C NMR (75 MHz, CD_3COCD_3) δ (ppm) 163.0 (C), 160.4 (2C), 138.3 (C), 136.6 (C), 130.5 (C), 128.7 (CH), 122.4 (CH) , 121.3 (CH), 120.4 (CH), 112.5 (CH), 109.6 (C), 109.4 (CH₂), 102.8 (CH), 91.8 (2CH), 56.3 (2CH₃), 55.7 (CH₃); HRMS (ESI) (M + H)⁺: m/z calcd for C₁₉H₂₀NO₃ 310.1443; found 310.1431.

 $1-(1-(4-Chlorophenyl)vinyl)-1H-indole$ 4b. Flash chromatography on silica gel (pentane) afforded 114 mg of 4b (0.45 mmol, yield 90%); white solid, mp: 69−70 °C; TLC: $R_f = 0.30$ (cyclohexane, SiO₂); IR (neat) 1491, 1455, 1338, 1215; ¹H NMR (300 MHz, C_6D_6) δ (ppm) 7.65 (dt, J = 7.7, 1.1 Hz, 1H), 7.12 (m, 3H), 6.92 (d, J = 8.7 Hz, 2H), 6.81 (d, J = 3.3 Hz, 1H), 6.76 (d, J = 8.7 Hz, 2H), 6.53 (d, J = 3.1 Hz, 1H), 5.03 (s, 1H), 4.93 (s, 1H); ¹³C NMR (75 MHz, C_6D_6) δ (ppm) 144.1 (C), 136.8 (C), 135.6 (C), 135.1 (C), 129.9 (C), 128.9 (2CH), 128.4 (2CH), 128.2 (CH), 122.5 (CH), 121.4 (CH), 120.8 (CH), 112.0 (CH), 108.2 (CH₂), 103.8 (CH); HRMS (APCI) $(M + H)^{+}$: m/ *z* calcd for $C_{16}H_{13}CIN$ 254.0737; found 254.0762.

Scheme 5. Cross-Coupling in the Presence of Deuterated Hydrazone d_2 -6

5-((tert-Butyldimethylsilyl)oxy)-1-(1-(2-chlorophenyl)vinyl)-1H-indole 4c. Flash chromatography on silica gel $(Et_2O/pentane 2/98)$ afforded 174 mg of 4c (0.46 mmol, yield 91%); colorless oil; TLC: R_f $= 0.71$ (EtOAc/cyclohexane, 10/90, SiO₂); IR (neat) 1739, 1571, 1467, 1294, 1194; ¹H NMR (300 MHz, C₆D₆) δ (ppm) 7.28 (d, J = 2.2 Hz, 1H), 7.19 (d, J = 0.5 Hz, 2H), 7.18−7.00 (m, 2H), 6.95−6.74 (m, 3H), 6.45 (dd, J = 3.3, 0.6 Hz, 1H), 5.26 (s, 1H), 4.97 (s, 1H), 1.08 (s, 9H), 0.20 (s, 6H); ¹³C NMR (75 MHz, C_6D_6) δ (ppm) 150.4 (C), 143.1 (C), 137.0 (C), 133.9 (C), 132.2 (C), 131.6 (CH), 131.4 (C), 130.5 (CH), 130.2 (CH), 128.4 (CH), 127.0 (CH), 116.8 (CH), 112.5(CH), 111.1 (CH), 107.8 (CH₂), 104.0 (CH), 26.0 (3CH₃), 18.5 (C), $-4.3(2CH_3)$; HRMS (APCI) (M + H)⁺: m/z calcd for $C_{22}H_{27}C$ INOSi 384.1550; found 384.1564.

6-Chloro-1-(1-(2-chlorophenyl)vinyl)-1H-indole 4d. Flash chromatography on silica gel (pentane) afforded 116 mg of 4d (0.41 mmol, yield 81%); colorless oil; TLC: $R_f = 0.26$ (cyclohexane, SiO₂); IR (neat) 1516, 1461, 1444, 1351, 1211; ¹H NMR (300 MHz, C_6D_6) δ (ppm) 7.43 (m, 1H), 7.30 (d, J = 8.4 Hz, 1H), 7.13 (m, 1H), 6.96 (m, 2H), 6.72 (m, 3H), 6.34 (dd, J = 3.4, 0.8 Hz, 1H), 5.07 (d, J = 0.7 Hz, 1H), 4.87 (d, J = 0.7 Hz, 1H); ¹³C NMR (75 MHz, C₆D₆) δ (ppm) 142.6 (C), 136.8 (C), 136.4 (C), 133.8 (C), 131.6 (CH), 130.5 (CH), 130.4 (CH), 129.0 (C), 128.7 (C), 128.4 (CH), 127.0 (CH), 122.4 (CH), 121.7 (CH), 112.1 (CH), 109.4 (CH₂), 104.3 (CH); HRMS $(APCI)$ $(M + H)^+$: m/z calcd for $C_{16}H_{12}Cl_2N$ 288.0347; found 288.0349.

5-Bromo-1-(1-(2,3,4-trichlorophenyl)vinyl)-1H-indole 4e. Flash chromatography on silica gel (pentane) afforded 170 mg of 4d (0.43 mmol, yield 85%); colorless oil; TLC: $R_f = 0.23$ (cyclohexane, SiO₂); IR (neat) 1518, 1453, 1367, 1207; ¹H NMR (300 MHz, CD₃COCD₃) δ (ppm) 7.78 (m, 1H), 7.73 (d, J = 8.4 Hz, 1H), 7.63 (d, J = 8.4 Hz, 1H), 7.24 (m, 2H), 7.17 (d, J = 8.8 Hz, 1H), 6.61 (dd, J = 3.4, 0.7 Hz, 1H), 5.75 (d, J = 1.2 Hz, 1H), 5.51 (d, J = 1.2 Hz, 1H); ¹³C NMR (75 MHz, CD_3COCD_3) δ (ppm) 142.5 (C), 137.8 (C), 135.4 (C), 135.3 (C), 132.5 (C), 131.4 (CH), 130.2 (C), 130.2 (CH), 130.0 (CH), 125.9 (CH), 124.3 (CH), 116.0 (C), 114.2 (C), 113.9 (CH), 111.3 (CH₂), 104.5 (CH); HRMS (APCI) $(M + H)^{+}$: m/z calcd for $C_{16}H_{10}BrCl_3N$ 399.9062; found 399.9082.

5-Bromo-1-(1-(3,4,5-trimethoxyphenyl)vinyl)-1H-indole 4f. Flash chromatography on silica gel (EtOAc/cyclohexane, 2/98) afforded 178 mg of 4f (0.46 mmol, yield 92%); colorless oil; TLC: $R_f = 0.26$ (EtOAc/cyclohexane, 20/80, SiO₂); IR (neat) 1736, 1581, 1504, 1449, 1412, 1370, 1335, 1230; ¹H NMR (300 MHz, CD₃COCD₃) δ (ppm) 7.79 (m, 1H), 7.37 (d, J = 3.3 Hz, 1H), 7.20 (dd, J = 8.8, 1.9 Hz, 1H), 7.06 (d, $J = 8.8$ Hz, 1H), 6.63 (dd, $J = 3.3$, 0.7 Hz, 1H), 6.61 (s, 2H), 5.72 (d, J = 0.6 Hz, 1H), 5.36 (d, J = 0.6 Hz, 1H), 3.76 (s, J = 11.2 Hz, 3H), 3.72 (s, 6H); ¹³C NMR (75 MHz, CD₃COCD₃) δ (ppm) 154.5 (2C), 145.4 (C), 140.5 (C), 136.1 (C), 133.0 (C), 132.1 (C), 131.2 (CH) , 125.4 (CH), 124.1 (CH), 114.3 (CH), 113.8 (C), 108.9 (CH₂), 105.4 (2CH), 103.4 (CH), 60.7 (CH₃), 56.5 (2CH₃); HRMS (ESI) $(M + H)^+$: m/z calcd for $C_{19}H_{19}NO_3Br$ 388.0548; found 388.0552.

1-(1-(2,3-Dihydrobenzo[b][1,4]dioxin-6-yl)vinyl)-5-fluoro-1H-indole 4g. Flash chromatography on silica gel (EtOAc/cyclohexane, 2/ 98) afforded 133 mg of 4g (0.45 mmol, yield 90%); white solid; mp: 129−131 °C; TLC: $R_f = 0.38$ (EtOAc/cyclohexane, 10/90, SiO₂); IR (neat) 1581, 1507, 1284, 1066; ¹H NMR (300 MHz, CD_3COCD_3) δ (ppm) 7.37 (d, J = 3.2 Hz, 1H), 7.32 (dd, J = 9.6, 2.5 Hz, 1H), 7.04 $(dd, J = 9.0, 4.5 Hz, 1H), 6.88 (dd, J = 9.2, 2.5 Hz, 1H), 6.83 (m, 1H),$ 6.75 (m, 2H), 6.62 (d, J = 3.2 Hz, 1H), 5.61 (s, 1H), 5.29 (s, 1H), 4.27 (m, 4H); ¹³C NMR (75 MHz, CD_3COCD_3) δ (ppm) 160.4 (C), 157.3 (C), 145.8 (C), 145.2 (C), 144.7 (C), 132.3 (C, d, J = 254 Hz), 131.6 (CH), 130.9 (C), 130.8 (C), 120.6 (CH), 118.2 (CH), 116.3 (CH), 113.5 (CH, d, $J = 10$ Hz), 110.7 (CH, d, $J = 26$ Hz), 108.1 (CH₂), 106.3 (CH, d, J = 24 Hz), 103.7 (CH, d, J = 4 Hz), 65.3 (CH₂), 65.1 (CH₂); HRMS (ESI) $(M + H)^{+}$: m/z calcd for $C_{18}H_{15}NO_2F$ 296.1087; found 296.1082.

5-Bromo-1-(1-(3-fluoro-4-methoxyphenyl)vinyl)-1H-indole 4h. Flash chromatography on silica gel (Et₂O/cyclohexane, 2/98) afforded 145 mg of 4h (0.42 mmol, yield 83%); white solid; mp: 83−84 °C; TLC: $R_f = 0.51$ (EtOAc/cyclohexane, 20/80, SiO₂); IR (neat) 1630, 1514, 1452, 1365, 1330, 1274, 1202; ¹H NMR (300 MHz, CDCl₃) δ (ppm) 7.79 (d, J = 1.7 Hz, 1H), 7.21 (m, 2H), 7.07 (dd, J = 12.1, 1.9 Hz, 1H), 7.95 (m, 3H), 6.58 (d, J = 3.2 Hz, 1H), 5.54 (s, 1H), 5.31 (s, 1H), 3.93 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ (ppm) 152.5 (C, d, $J = 247$ Hz), 148.8 (C, d, $J = 11$ Hz), 143.6 (C), 135.2 (C), 131.1 (C), 129.9 (CH), 129.8 (C), 125.1 (CH), 123.6 (CH), 123.0 (CH, d, J = 3.1 Hz), 114.8 (CH, d, J = 20 Hz), 113.7 (C), 113.4 (2CH), 108.1 $(CH₂)$, 102.9 (CH), 56.4 (CH₃); HRMS (ESI) (M + Na)⁺: m/z calcd for $C_{17}H_{13}BrFNNaO$ 368.0062; found 368.0060.

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3-(1-(5-Fluoro-1H-indol-1-yl)vinyl)-9-methyl-9H-carbazole 4i. Flash chromatography on silica gel (EtOAc/cyclohexane, 2/98) afforded 151 mg of 4i (0.45 mmol, yield 89%); white solid; mp: 137−139 °C; TLC: $R_f = 0.56$ (EtOAc/cyclohexane, 20/80, SiO₂); IR (neat) 2927, 1601, 1469, 1446, 1369, 1247, 1188; ¹ H NMR (300 MHz, CD_3COCD_3) δ (ppm) 8.16 (dd, J = 1.8, 0.7 Hz, 1H), 8.08 (dt, J = 7.8, 1.0 Hz, 1H), 7.67−7.41 (m, 4H), 7.35 (ddd, J = 8.6, 3.9, 2.2 Hz, $2H$), 7.20 (ddd, J = 8.0, 6.9, 1.2 Hz, 1H), 7.01 (ddt, J = 9.0, 4.5, 0.7 Hz, 1H), 6.87−6.72 (m, 1H), 6.66 (dd, J = 3.3, 0.8 Hz, 1H), 5.67 (s, 1H), 5.34 (s, 1H), 3.91 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ (ppm) 158.2 (C, d, J = 235 Hz), 146.0 (C), 141.7 (C), 133.3 (C), 130.6 (CH), 130.0 (C), 129.8 (C), 128.0 (C), 126.4 (CH), 125.1 (CH), 123.1 (C), 122.8 (C), 120.7 (CH), 119.5 (CH), 119.3 (CH), 113.0 $(CH, d, J = 9 Hz)$, 110.5 (CH, d, $J = 21 Hz$), 108.9 (CH), 108.6 (CH), 106.3 (CH₂), 105.8 (d, J = 23 Hz), 103.0 (CH, d, J = 4 Hz), 29.2 (CH₃); HRMS (APCI) (M + H)⁺: m/z calcd for C₂₃H₁₈FN₂ 341.1454; found 341.1446.

1-(1-(Naphthalen-2-yl)vinyl)-1H-benzo[g]indole 4j. Flash chromatography on silica gel (pentane) afforded 100 mg of 4j (0.31 mmol, yield 62%); white solid; mp: 105−107 °C; TLC: $R_f = 0.20$ (cyclohexane, SiO₂); IR (neat) 1737, 1498, 1443, 1400, 1348, 1328, 1232; ¹H NMR (300 MHz, C_6D_6) δ (ppm) 8.41 (m, 1H), 7.78 (m, 1H), 7.56 (d, J = 8.6 Hz, 1H), 7.30 (dd, J = 8.7, 1.8 Hz, 1H), 7.11 (m, 10H), 6.80 (d, J = 3.0 Hz, 1H), 5.70 (s, 1H), 5.24 (s, 1H); 13C NMR (75 MHz, C_6D_6) δ (ppm) 147.7 (C), 134.4 (C), 134.0 (C), 133.8 (C), 132.1 (C), 130.9 (C), 129.7 (CH), 129.4 (CH), 129.1 (CH), 129.0 (CH), 126.9 (CH), 126.6 (CH), 126.5 (C), 125.8 (CH), 125.4 (CH), 123.9 (CH), 123.3 (CH), 122.6 (CH), 121.8 (CH), 121.4 (CH), 113.0 (CH₂), 104.6 (CH); HRMS (APCI) $(M + H)^{+}$: *m/z* calcd for $C_{24}H_{18}N$ 320.1439; found 320.1451.

4-(1-(5-Methyl-1H-indol-1-yl)vinyl)aniline 4k. Flash chromatography on silica gel (DCM/cyclohexane, 40/60) afforded 87 mg of 4k (0.35 mmol, yield 70%); colorless oil; TLC: $R_f = 0.30$ (EtOAc/ cyclohexane, $30/70$, $SiO₂$); IR (neat) 1621, 1515, 1475, 1365, 1331; ¹H NMR (300 MHz, MeOD) δ (ppm) 7.34 (m, 1H), 7.17 (d, J = 3.2 Hz, 1H), 7.00 (d, $J = 8.8$ Hz, 2H), 6.92 (d, $J = 8.4$ Hz, 1H), 6.84 (dd, J $= 8.5, 1.5$ Hz, 1H), 6.64 (d, J = 8.7 Hz, 2H), 6.46 (dd, J = 3.3, 0.7 Hz, 1H), 5.35 (s, 1H), 5.06 (s, 1H), 2.38 (s, 3H); 13C NMR (75 MHz, MeOD) δ 150.3 (C), 147.1 (C), 136.3 (C), 131.1 (C), 130.1 (C), 129.8 (CH), 129.0 (2CH), 127.6 (C), 124.2 (CH), 121.3 (CH), 115.8 $(2CH)$, 112.7 (CH), 104.2 (CH₂), 103.1 (CH), 21.4 (CH₃); HRMS (ESI) $(M + H)^{+}$: m/z calcd for $C_{17}H_{17}N_2$ 249.1392; found 249.1390.

1-(1-(4-(Phenylethynyl)phenyl)vinyl)-1H-indole-4-carbonitrile 4l. Flash chromatography on silica gel (EtOAc/cyclohexane, 2/98) afforded 124 mg of 4l (0.36 mmol, yield 72%); white solid; mp: 113−115 °C; TLC: R_f = 0.52 (EtOAc/cyclohexane, 20/80, SiO₂); IR (neat) 2223, 1628, 1510, 1432, 1332; ¹H NMR (300 MHz, C_6D_6) δ 7.52 (m, 2H), 7.35 (d, J = 8.6 Hz, 2H), 7.12 (dd, J = 7.4, 0.8 Hz, 1H), 7.00 (m, 3H), 6.96 (d, J = 8.4 Hz, 1H), 6.78 (d, J = 8.5 Hz, 2H), 6.73 $(dd, J = 3.3, 0.8$ Hz, 1H), 6.73 (m, 2H), 5.09 (d, $J = 0.6$ Hz, 1H), 4.77 (d, J = 0.6 Hz, 1H); ¹³C NMR (75 MHz, C₆D₆) δ 143.8 (C), 136.3 (C), 136.1 (C), 132.3 (2CH), 132.0 (2CH), 131.1 (C),131.1 (CH), 128.9 (CH), 128.8 (2CH), 126.9 (2CH), 125.7 (CH), 125.0 (CH), 123.5 (C), 122.1 (C), 118.4 (CH), 116.1 (C), 110.0 (CH₂), 104.6 (C), 102.5 (CH), 92.0 (C), 89.4 (C); HRMS (ESI) (M + H)⁺: m/z calcd for $C_{25}H_{17}N_2$ 345.1392; found 345.1390.

1-(1-(2,4-Dimethoxyphenyl)prop-1-en-1-yl)-1H-indole-4-carbonitrile 4m. Flash chromatography on silica gel (EtOAc/cyclohexane, 2/ 98) afforded 125 mg of 4m (0.39 mmol, yield 78%); yellow oil; TLC: $R_f = 0.42$ (EtOAc/cyclohexane, 10/90, SiO₂); IR (neat) 2225, 1736, 1609, 1576, 1503, 1433, 1337, 1303, 1240, 1208. Data for Z major isomer: ¹H NMR (400 MHz, CD_3COCD_3) δ (ppm) 7.50 (dd, J = 7.3, 0.7 Hz, 1H), 7.46 (d, $J = 3.2$ Hz, 1H), 7.40 (d, $J = 8.3$ Hz, 1H), 7.23 $(m, 1H)$, 6.92 (d, J = 8.5 Hz, 1H), 6.75 (dd, J = 3.2, 0.7 Hz, 1H), 6.53 $(d, J = 2.4 \text{ Hz}, 1H)$, 6.46 $(dd, J = 8.5, 2.4 \text{ Hz}, 1H)$, 6.32 $(q, J = 7.0 \text{ Hz},$ 1H), 3.78 (s, 3H), 3.59 (s, 3H), 1.58 (d, J = 7.0 Hz, 3H). Data for Z major isomer: ¹³C NMR (101 MHz, CD_3COCD_3) δ 162.2 (C), 159.5 (C), 137.1 (C), 135.2 (C), 133.0 (CH), 131.0 (CH), 130.2 (C), 125.9 (CH), 125.6 (CH), 122.3 (CH), 120.5 (C), 119.0 (C), 117.0 (CH), 105.7 (CH), 103.6 (C), 101.0 (CH), 99.6 (CH), 55.9 (CH₃), 55.7

(CH₃), 13.8 (CH₃); HRMS (ESI) $(M + H)^{+}$: m/z calcd for $C_{20}H_{19}N_2O_2$ 319.1447; found 319.1443.

1-(3,3-Dimethylbut-1-en-2-yl)-5-methoxy-1H-indole 4n. Flash chromatography on silica gel (cyclohexane) afforded 93 mg of 4n (0.41 mmol, yield 81%); colorless oil; TLC: $R_f = 0.20$ (cyclohexane, $SiO₂$); IR (neat) 1639, 1478, 1447, 1288, 1269, 1209, 1162, 1146; ¹H NMR (300 MHz, CDCl3) δ (ppm) 7.21 (d, J = 8.9 Hz, 1H), 7.09 (m, 2H), 6.86 (dd, $J = 8.9$, 2.5 Hz, 1H), 6.50 (d, $J = 3.1$ Hz, 1H), 5.52 (s, 1H), 5.14 (s, 1H), 3.88 (s, 3H), 1.20 (s, 9H); 13C NMR (75 MHz, CDCl3) δ (ppm) 154.5 (C), 154.1 (C), 133.6 (C), 129.5 (CH), $128.1(C)$, 113.4 (CH₂), 112.2 (CH), 112.0 (CH), 102.1 (CH), 101.0 (CH) , 56.0 (CH_3) , 38.0 (C) , 29.5 $(3CH_3)$; HRMS $(APCI)$ $(M + H)^+$: m/z calcd for C₁₅H₂₀NO 230.1545; found 230.1541.

5-Methoxy-1-(pent-2-en-3-yl)-1H-indole 40. Flash chromatography on silica gel (EtOAc/cyclohexane, 5/95) afforded 77 mg of 4o (0.36 mmol, yield 72%); colorless oil; TLC: $R_f = 0.28$ (EtOAc/ cyclohexane, 2/98, SiO₂); IR (neat) 1476, 1436, 1239, 1215, 1193, 1167, 1147; Data for Z and E isomers: ¹H NMR (300 MHz, C_6D_6) δ (ppm) 7.28 (d, $J = 8.8$ Hz, 1H), 7.16 (m, 2H), 7.12 (d, $J = 8.8$ Hz, 1H), 7.11 (d, J = 8.8 Hz, 1H), 7.03 (d, J = 8.8 Hz, 1H), 6.91 (d, J = 3.1) Hz, 1H), 6.81 (d, J = 3.1 Hz, 1H), 6.58 (dd, J = 3.1, 0.7 Hz, 1H), 6.55 $(dd, J = 3.1, 0.7 Hz, 1H), 5.39 (q, J = 7.0 Hz, 1H), 5.29 (qt, J = 6.8, 1.2)$ Hz, 1H), 3.55 (s, 3H), 3.55 (s, 3H), 2.24 (q, J = 7.5 Hz, 2H), 2.15 (qt, $J = 7.4$, 1.3 Hz, 2H), 1.46 (d, $J = 7.0$ Hz, 3H), 1.25 (dt, $J = 6.8$, 1.4 Hz, 3H), 0.73 (t, J = 7.3 Hz, 3H), 0.68 (t, J = 7.3 Hz, 3H). Data for Z and E isomers: ¹³C NMR (75 MHz, C_6D_6) δ 155.0 (C), 155.0 (C), 139.8 (C), 139.7 (C), 129.7 (C), 129.1 (C), 127.4 (C), 127.3 (C), 120.0 (CH), 119.4 (CH), 112.8 (CH), 112.7 (CH), 111.8 (CH), 111.6 (CH), 102.8 (CH), 102.7 (CH), 102.4 (CH), 102.0 (CH), 55.3 $(CH₃)$, 55.3 (CH₃), 29.9 (CH₂), 23.5 (CH₂), 12.7 (CH₃), 12.4 (CH₃), 12.0 (CH₃), 11.7 (CH₃); HRMS (ESI) $(M + H)^{+}$: m/z calcd for $C_{14}H_{18}NO$ 216.1388; found 216.1380.

 $1-(Cyclohexylidenemethyl)-5-methoxy-1H-indole$ 4p. Flash chromatography on silica gel (cyclohexane) afforded 72 mg of 4p (0.30 mmol, yield 60%); colorless oil; TLC: $R_f = 0.33$ (EtOAc/cyclohexane, $(5/95, SiO₂)$; IR (neat) 1605, 1510, 1442, 1244, 1176; ¹H NMR (300 MHz, CD_3COCD_3) δ 7.19 (d, J = 8.8 Hz, 1H), 7.10 (d, J = 13.5 Hz, 1H), 7.09 (d, $J = 12.7$ Hz, 1H), 6.81 (dd, $J = 8.8$, 2.4 Hz, 1H), 6.62 (s, 1H), 6.44 (dd, J = 3.1, 0.8 Hz, 1H), 3.80 (s, 3H), 2.33 (m, 2H), 2.17 (m, 2H), 1.65 (m, 4H), 1.53 (m, 2H); 13C NMR (75 MHz, CD3COCD3) δ 155.5 (C), 140.2 (2C), 132.9 (C), 129.7 (CH), 117.8 (CH), 112.7 (CH), 111.6 (CH), 103.2 (CH), 102.4 (CH), 55.9 (CH_3) , 33.9 (CH_2) , 29.0 $(2CH_2)$, 28.1 (CH_2) , 27.1 (CH_2) ; HRMS (ESI) $(M + H)^{+}$: m/z calcd for $C_{16}H_{20}NO$ 242.1545; found 242.1543.

1-(Cyclopentylidene(3,4,5-trimethoxyphenyl)methyl)-5-fluoro-1H-indole 4q. Flash chromatography on silica gel (EtOAc/cyclohexane, 2/98) afforded 149 mg of 4q (0.39 mmol, yield 78%); white solid; mp: 125−127 °C; TLC: R_f = 0.37 (EtOAc/cyclohexane, 20/80, $SiO₂$); IR (neat) 1581, 1506, 1474, 1448, 1413, 1341, 1237, 1128; ¹H NMR (300 MHz, C_6D_6) δ 7.39 (dd, J = 9.3, 2.3 Hz, 1H), 7.05 (m, 1H), 6.96 (m, 2H), 6.49 (dd, J = 3.1, 0.7 Hz, 1H), 6.41 (s, 2H), 3.80 $(s, 3H)$, 3.29 $(s, 6H)$, 2.51 $(t, J = 7.0 \text{ Hz}, 2H)$, 2.03 $(t, J = 7.2 \text{ Hz}, 2H)$, 1.51 (m, 2H), 1.36 (m, 2H); ¹³C NMR (75 MHz, C_6D_6) δ 158.82 (C, d, $J = 234$ Hz), 154.0 (2C), 144.8 (C), 134.1 (C), 133.2 (C), 130.1 (CH), 129.1 (C, d, $J = 10$ Hz), 128.7 (C), 111.7 (CH, d, $J = 10$ Hz), 110.9 (CH, d, J = 26 Hz), 106.3 (CH, d, J = 23 Hz), 105.9 (2CH), 102.8 (C, d, J = 5 Hz), 60.5 (CH₃), 55.9 (2CH₃), 33.1 (CH₂), 32.4 $(CH₂)$, 27.9 (CH₂), 26.1 (CH₂); HRMS (ESI) (M + Na)⁺: m/z calcd for $C_{23}H_{24}FNNaO_3$ 404.1638; found 404.1615.

5-Chloro-1-(cyclobutylidene(4-fluorophenyl)methyl)-1H-indole 4r. Flash chromatography on silica gel (pentane) afforded 130 mg of 4r (0.42 mmol, yield 84%); white solid; mp: 71–74 °C; TLC: R_f = 0.47 (cyclohexane, SiO₂); IR (neat) 1602, 1507, 1454, 1370, 1328, 1231, 1208, 1159; ¹H NMR (300 MHz, C₆D₆) δ (ppm) 7.68 (m, 1H), 7.17 (m, 1H), 6.86 (d, J = 8.7 Hz, 1H), 6.74 (d, J = 3.2 Hz, 1H), 6.69 $(m, 4H)$, 6.39 (dd, J = 3.2, 0.8 Hz, 1H), 2.66 $(m, 2H)$, 2.30 $(m, 2H)$, 1.64 (m, 2H). ¹³C NMR (75 MHz, C₆D₆) δ (ppm) 162.3 (C, d, J = 247 Hz), 142.0 (C), 135.0 (C), 133.0 (C), 132.9 (C), 129.9 (C), 129.6 (CH), 128.2 (2CH), 126.3 (C), 122.9 (CH), 121.0 (CH), 115.7 $(2CH, d, J = 21.6 Hz)$, 112.1 (CH), 102.7 (CH), 31.6 (CH₂), 30.4

The Journal of Organic Chemistry **Article Article Article Article Article Article Article Article Article**

 (CH_2) , 17.4 (CH_2) ; HRMS (ESI) $(M + H)^+$: m/z calcd for $C_{19}H_{16}$ ClFN 312.0955; found 312.0939.

(Z)-1-(1,2-Diphenylvinyl)-5-methyl-1H-indole 4s. Flash chromatography on silica gel (pentane) afforded 124 mg of 4s (0.40 mmol, yield 80%); white solid; mp: 100-102 °C; TLC: $R_f = 0.33$ (cyclohexane, SiO_2); IR (neat) 1474, 1448, 1391, 1211; ¹H NMR $(300 \text{ MHz}, \text{C}_6\text{D}_6)$ δ 7.45 (m, 1H), 7.12 (m, 2H), 7.02 (m, 4H), 6.84 $(m, 7H)$, 6.78 (d, J = 3.2 Hz, 1H), 6.58 (dd, J = 3.2, 0.8 Hz, 1H), 2.30 (s, 3H); ¹³C NMR (75 MHz, C_6D_6) δ 139.0 (C), 136.9 (C), 135.4 (C), 134.6 (C), 129.9 (C), 129.8 (C), 129.1 (2CH), 128.9 (2CH), 128.7 (2CH), 126.6 (2CH), 125.0 (CH), 124.5 (CH), 121.2 (CH), 111.9 (CH), 104.0 (CH), 21.5 (CH₃); HRMS (ESI) (M + H)⁺: m/z calcd for $C_{23}H_{20}N$ 310.1596; found 310.1568.

(Z)-1-(1,2-Diphenylvinyl)-5-methyl-1H-indole 4t. Flash chromatography on silica gel (EtOAc/cyclohexane, 1/99) afforded 160 mg of 4t (0.45 mmol, yield 90%); white solid; mp: 109−100 °C; TLC: Rf = 0.39 (EtOAc/cyclohexane, 10/90, SiO₂); IR (neat) 1735, 1606, 1512, 1456, 1301, 1244, 1216, 1175; ¹H NMR (300 MHz, CD₃COCD₃) δ (ppm) 7.84−7.51 (m, 1H), 7.22−6.94 (m, 7H), 6.91−6.82 (m, 2H), 6.79−6.70 (m, 3H), 6.64 (d, J = 9.0 Hz, 2H), 3.79 (s, 3H), 3.69 (s, 3H); ¹³C NMR (75 MHz, CD₃COCD₃) δ (ppm).161.0 (C), 160.1 (C), 136.4 (C), 134.4 (C), 132.0 (C), 130.9 (2CH), 129.8 (C), 129.3 (CH), 128.5(C), 127.7 (2CH), 124.1 (CH), 122.9 (CH), 121.6 (CH), 120.9 (CH), 114.9 (2CH), 114.5 (2CH), 112.1 (CH), 104.3 (CH), 55.7 (CH₃), 55.4 (CH₃); HRMS (ESI) $(M + H)^{+}$: m/z calcd for $C_{24}H_{22}NO_2$ 356.1651; found 356.1649.

1-(2H-Chromen-4-yl)-5-methyl-1H-indole 4u. Flash chromatography on silica gel (Et₂O/cyclohexane, 1/99) afforded 94 mg of 4u (0.36) mmol, yield 72%); colorless oil; TLC: $R_f = 0.56$ (EtOAc/cyclohexane, $(5/95, SiO₂)$; IR (neat) 1648, 1484, 1385, 1335, 1224, 1173, 1121; ¹H NMR (300 MHz, CD_3COCD_3) δ (ppm) 7.42 (m, 1H), 7.26 (d, J = 3.2 Hz, 1H), 7.22 (td, $J = 7.8$, 1.6 Hz, 1H), 7.09 (d, $J = 8.4$ Hz, 1H), 6.93 (m, 2H), 6.82 (td, $J = 7.6$, 1.1 Hz, 1H), 6.57 (m, 2H), 6.04 (t, $J =$ 3.9 Hz, 1H), 5.04 (d, $J = 3.9$ Hz, 2H), 2.40 (s, 3H); ¹³C NMR (75 MHz, CD_3COCD_3) δ (ppm) 156.2 (C), 154.0 (C), 134.4 (C), 131.1 (CH), 130.2 (C), 130.0 (C), 129.3 (CH), 124.4 (CH), 124.4 (CH), 122.2 (CH), 122.0 (C), 121.4 (CH), 118.8 (CH), 117.1 (CH), 111.6 (CH), 103.4 (CH), 66.0 (CH₂), 21.4 (CH₃); HRMS (APCI) (M + H)⁺: m/z calcd for C₁₈H₁₆NO 262.1232; found 262.1239.

5-Methoxy-1-(6-methoxy-3,4-dihydronaphthalen-1-yl)-1H-indole 4v. Flash chromatography on silica gel (EtOAc/cyclohexane, 2/98) afforded 131 mg of 4v (0.43 mmol, yield 86%); colorless oil; TLC: R_f = 0.44 (EtOAc/cyclohexane, 20/80, SiO₂); IR (neat) 3391, 3293, 1604, 1476, 1280, 1252; ¹H NMR (300 MHz, C₆D₆) δ 7.18 (d, J = 2.3 Hz, 1H), 7.13 (m, 1H), 7.06 (d, J = 3.1 Hz, 1H), 7.00 (dd, J = 8.9, 2.3 Hz, 1H), 6.69 (d, $J = 2.6$ Hz, 1H), 6.62 (m, 2H), 6.30 (dd, $J = 8.5$, 2.6 Hz, 1H), 5.65 (t, J = 4.7 Hz, 1H), 3.52 (s, 3H), 3.24 (s, 3H), 2.54 (t, J $= 7.9$ Hz, 2H), 2.11–1.93 (m, 2H); ¹³C NMR (75 MHz, C₆D₆) δ 160.1 (C), 155.3 (C), 138.4 (C), 136.9 (C), 132.7 (C), 129.9 (C), 129.2 (CH), 125.6 (C), 125.2 (CH), 121.7 (CH), 114.7 (CH), 112.9 (CH), 112.6 (CH), 111.2 (CH), 103.0 (CH), 102.6 (CH), 55.5 (CH_3) , 54.8 (CH₃), 28.3 (CH₂), 22.9 (CH₂); HRMS (ESI) (M + H)⁺: m/z calcd for $C_{20}H_{20}NO_2$ 306.1494; found 306.1499.

2-Phenyl-1-(1-(2,4,6-trimethoxyphenyl)vinyl)-1H-indole 4w. Flash chromatography on silica gel (EtOAc/cyclohexane 2/98) afforded 182 mg of 4w (0.48 mmol, yield 95%); white solid ; mp: 114−117 °C; TLC: $R_f = 0.61$ (EtOAc/cyclohexane, 20/80, SiO₂); IR (neat) 1603, 1581, 1454, 1413, 1332, 1226, 1204; ¹H NMR (300 MHz, CDCl₃) δ (ppm) 7.62 (d, J = 7.2 Hz, 1H), 7.49 (m, 3H), 7.19 (m, 5H), 6.55 (s, 1H), 5.86 (s, 2H), 5.66 (s, 1H), 5.59 (s, 1H), 3.73 (s, 3H), 3.38 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ (ppm) 161.2 (C), 159.3 (2C), 141.2 (C), 139.7 (C), 135.5 (C), 133.9 (C), 128.9 (2CH), 128.4 (C), 127.5 (2CH), 126.8 (CH), 121.7 (CH), 120.1 (CH), 119.9 (CH), 116.6 (CH₂), 111.8 (CH), 109.7 (C), 103.5 (CH), 90.5 (2CH), 55.6 $(2CH_3)$, 55.4 (CH_3) ; HRMS (ESI) $(M + Na)^+$: m/z calcd for $C_{25}H_{23}NNaO_3$ 408.1576; found 408.1549.

3-Methyl-1-(1-(2,4,6-trimethoxyphenyl)vinyl)-1H-indole 4x. Flash chromatography on silica gel (EtOAc/cyclohexane 2/98) afforded 149 mg of 4x (0.46 mmol, yield 92%); white solid; mp: 134−136 °C; TLC: $R_f = 0.35$ (EtOAc/cyclohexane, 10/90, SiO₂); IR (neat) 1604, 1582,

1496, 1453, 1414, 1355, 1225, 1204; ¹ H NMR (300 MHz, CD_3COCD_3) δ (ppm) 7.47 (m, 1H), 7.32 (m, 1H), 7.03 (m, 2H), 6.94 (d, J = 1.1 Hz, 1H), 6.29 (s, 2H), 5.49 (s, 1H), 4.96 (s, 1H), 3.85 (s, 3H), 3.66 (s, 6H), 2.24 (d, $J = 1.1$ Hz, 3H); ¹³C NMR (75 MHz, CD₃COCD₃</sub>) δ (ppm) 163.0 (C), 160.4 (2C), 138.2 (C), 136.9 (C), 130.9 (C), 129.7 (C), 126.1 (CH), 122.5 (CH), 119.9 (CH), 119.4 (CH), 112.5 (CH), 111.7 (C), 107.6 (CH₂), 91.8 (2CH), 56.3 $(2CH₃)$, 55.8 (CH₃), 9.6 (CH₃); HRMS (ESI) (M + H)⁺: m/z calcd for $C_{20}H_{22}NO_3$ 324.1600; found 324.1594.

(Z)-Ethyl 1-(1,2-Diphenylvinyl)-1H-indole-2-carboxylate 4y. Flash chromatography on silica gel (EtOAc/cyclohexane 1/99) afforded 114 mg of 4y (0.31 mmol, yield 62%); white solid ; mp: 124−126 °C; TLC: $R_f = 0.57$ (EtOAc/cyclohexane, 10/90, SiO₂); IR (neat) 1712, 1524, 1448, 1406, 1237, 1210, 1193; ¹H NMR (300 MHz, C₆D₆) δ (ppm) 7.61 (m, 2H), 7.17 (m, 4H), 7.01 (m, 5H), 6.91 (m, 2H), 6.81 $(m, 3H)$, 3.85 (qq, J = 10.8, 7.1 Hz, 2H), 0.80 (t, J = 7.1 Hz, 3H); ¹³C NMR (75 MHz, C₆D₆) δ (ppm) 160.6 (C), 139.8 (C), 138.9 (C), 136.1 (C), 135.5 (C), 129.6 (C), 128.9 (2CH), 128.8 (2CH), 128.8 (2CH), 128.5 (CH), 127.3 (C), 126.8 (CH), 126.1 (CH), 125.8 (2CH), 122.8 (CH), 122.0 (CH), 112.3 (CH), 112.2 (CH), 60.5 $(CH₂)$, 14.0 $(CH₃)$; HRMS (ESI) $(M + Na)^+$: m/z calcd for $C_{25}H_{21}NNaO_2$ 390.1470; found 390.1471.

1-(1-(2,4,6-Trimethoxyphenyl)vinyl)-1H-indole-3-carbonitrile 4z. Flash chromatography on silica gel (EtOAc/cyclohexane 5/95) afforded 110 mg of 4z (0.33 mmol, yield 65%); white solid; mp: 116−118 °C; TLC: $R_f = 0.33$ (EtOAc/cyclohexane, 20/80, SiO₂); IR (neat) 1647, 1581, 1496, 1314, 1226, 1205, 1158; ¹H NMR (300 MHz, CDCl3) δ (ppm) 7.77 (m, 1H), 7.69 (s, 1H), 7.52 (m, 1H), 7.30 (m, 2H), 6.19 (s, 2H), 5.76 (s, 1H), 5.46 (s, 1H), 3.89 (s, 3H), 3.73 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ (ppm) 162.4 (C), 159.5 (2C), 136.1 (C), 135.5 (C), 135.3 (CH), 128.1 (C), 123.9 (CH), 122.2 (CH), 119.7 (CH), 116.2 (C), 114.0 (CH₂), 112.7 (CH), 107.3 (C), 90.9 (2CH), 86.5 (C), 56.0 (2CH₃), 55.6 (CH₃); HRMS (ESI) (M + Na)⁺: m/z calcd for $C_{20}H_{18}N_2NaO_3$ 357.1215; found 357.1187.

2-(1-(1-(2-Chlorophenyl)vinyl)-1H-indol-3-yl)ethanamine 4aa. Flash chromatography on silica gel (MeOH/DCM 2/98) afforded 115 mg of 4aa (0.39 mmol, yield 78%); yellow oil ; TLC: $R_f = 0.47$ (MeOH/DCM, 10/90, SiO₂); IR (neat) 3467, 1735, 1455, 1376, 1225, 1046; ¹ H NMR (300 MHz, MeOD) δ (ppm) 7.55 (m, 2H), 7.38 (m, 3H), 7.06 (m, 3H), 6.95 (s, 1H), 5.53 (d, J = 0.5 Hz, 1H), 5.25 (d, $J = 0.4$ Hz, 1H), 2.90 (m, 4H); ¹³C NMR (75 MHz, MeOD) δ (ppm) 144.5 (C), 137.9 (C), 137.6 (C), 134.5 (C), 132.8 (CH), 131.6 (CH), 131.2 (CH), 130.6 (C), 128.4 (CH), 126.7 (CH), 123.4 (CH) , 121.1 (CH), 119.9 (CH), 115.7 (C), 112.6 (CH), 108.3 (CH₂), 42.5 (CH₂), 28.9 (CH₂); HRMS (ESI) $(M + H)^{+}$: m/z calcd for $C_{18}H_{18}CIN_2$ 297.1159; found 297.1139.

2-(1-(1-(2,4,6-Trimethoxyphenyl)vinyl)-1H-indol-3-yl)acetonitrile 4ab. Flash chromatography on silica gel (EtOAc/cyclohexane 5/95) afforded 143 mg of 4ab (0.41 mmol, yield 82%); yellow solid; mp: 161−163 °C; TLC: R_f = 0.27 (EtOAc/cyclohexane, 20/80, SiO₂); IR (neat) 1641, 1604, 1582, 1455, 1415, 1366, 1226; ¹H NMR (300 MHz, CD_3COCD_3) δ 7.61 (m, 1H), 7.41 (m, 1H), 7.23 (s, 1H), 7.12 $(m, 2H)$, 6.29 (s, 2H), 5.59 (s, 1H), 5.13 (s, 1H), 3.97 (d, J = 0.9 Hz, 3H), 3.85 (s, 3H), 3.67 (s, 6H); ¹³C NMR (75 MHz, CD₃COCD₃) δ 163.2 (C), 160.4 (2C), 137.9 (C), 137.1 (C), 128.6 (C), 127.4 (C), 127.4 (CH), 123.3 (CH), 120.7 (CH), 119.2 (CH), 112.9 (CH), 110.1 (CH₂), 109.2 (C), 106.0 (C), 91.8 (2CH), 56.3 (2CH₃), 55.8 (CH₃), 14.0 (CH₂); HRMS (ESI) $(M + H)^{+}$: m/z calcd for $C_{21}H_{21}N_2O_3$ 349,1552; found 349.1553.

3-(1-(6-Methoxy-3,4-dihydronaphthalen-1-yl)-1H-indol-3-yl)-1 methylquinoxalin-2(1H)-one 4ac. Flash chromatography on silica gel (DCM/cyclohexane 50/50) afforded 195 mg of 4ac (0.45 mmol, yield 90%); yellow solid; mp: 190−191 °C; TLC: $R_f = 0.61$ (DCM, SiO₂); IR (neat) 1738, 1650, 1535, 1497, 1455, 1307, 1253, 1200, 1167; ¹H NMR (300 MHz, CDCl₃) δ (ppm) 9.09 (d, J = 8.0 Hz, 1H), 9.00 (s, 1H), 8.07 (d, J = 7.9 Hz, 1H), 7.51 (m, 1H), 7.38 (m, 3H), 7.25 (m, 2H), 6.83 (s, 1H), 6.57 (m, 2H), 6.20 (t, J = 4.6 Hz, 1H), 3.82 (s, 3H), 3.80 (s, 3H), 2.99 (t, $J = 7.8$ Hz, 2H), 2.60 (m, 2H); ¹³C NMR (75 MHz, CDCl₃) δ (ppm) 159.7 (C), 154.7 (C), 151.0 (C), 138.2 (C), 137.4 (C), 136.6 (CH), 136.0 (C), 134.0 (C), 131.9 (C), 129.6 (CH), 128.5(CH), 127.5 (C), 124.8 (C), 124.7 (CH), 123.7 (CH), 123.7 (CH), 123.5 (CH), 123.2 (CH), 122.1 (CH), 114.3 (CH), 113.6 (CH), 112.6 (C), 111.6 (CH), 111.2 (CH), 55.4 (CH₃), 29.3 (CH₃), 28.2 (CH₂), 22.9 (CH₂); HRMS (ESI) $(M + H)^{+}$: m/z calcd for $C_{28}H_{24}N_2O_2$ 434.1869; found 434.1868.

1-(2-(1-(2H-Chromen-4-yl)-1H-indol-3-yl)ethyl)piperidin-2-one 4ad. Flash chromatography on silica gel (MeOH/DCM 2/98) afforded 145 mg of 4ad (0.39 mmol, yield 78%); colorless oil; TLC: $R_f = 0.67$ (MeOH/DCM, 5/95, SiO₂); IR (neat) 1736, 1640, 1606, 1486, 1459, 1395, 1224; ¹H NMR (300 MHz, CD_3COCD_3) δ 7.75 $(m, 1H)$, 7.17 $(m, 5H)$, 6.91 (dd, J = 8.1, 0.9 Hz, 1H), 6.82 (td, J = 7.6, 1.1 Hz, 1H), 6.60 (dd, $J = 7.7$, 1.6 Hz, 1H), 6.03 (t, $J = 3.9$ Hz, 1H), 5.03 (d, J = 3.9 Hz, 2H), 3.65 (m, 2H), 3.27 (m, 2H), 3.04 (m, 2H), 2.24 (m, 2H), 1.71 (m, 4H); ¹³C NMR (75 MHz, CD_3COCD_3) δ 169.2 (C), 156.2 (C), 137.9 (C), 134.3 (C), 131.1 (CH), 129.6 (C), 126.9 (CH), 124.4 (CH), 123.0 (CH), 122.2 (CH), 122.0 (C), 120.6 (CH), 120.1 (CH), 118.7 (CH), 117.1 (CH), 115.2 (C), 111.8 (CH), 66.0 (CH₂), 49.0 (CH₂), 48.5 (CH₂), 33.1 (CH₂), 24.1 (CH₂), 23.6 (CH₂), 22.2 (CH₂); HRMS (ESI) $(M + Na)^{+}$: m/z calcd for C24H24N2O2Na 395.1735; found 395.1728.

2,3-Diphenyl-1-(1-(3,4,5-trimethoxyphenyl)vinyl)-1H-indole 4ae. Flash chromatography on silica gel (EtOAc/cyclohexane 1/99) afforded 203 mg of 4ae (0.44 mmol, yield 88%); white solid ; mp: 149−151 °C; TLC: R_f = 0.40 (EtOAc/cyclohexane, 20/80, SiO₂); IR (neat) 1581, 1504, 1454, 1413, 1368, 1323, 1238; ¹ H NMR (300 MHz, CD_3COCD_3) δ 7.73 (m, 1H), 7.34 (m, 7H), 7.21 (m, 6H), 6.37 (s, 2H), 6.04 (s, 1H), 5.42 (s, 1H), 3.67 (s, 3H), 3.63 (s, 6H); 13C NMR (75 MHz, CD_3COCD_3) δ 154.3 (2C), 144.0 (C), 139.1 (C), 138.7 (C), 135.8 (C), 133.8 (C), 133.0 (C), 131.8 (2CH), 130.9 (2CH), 129.2 (2CH), 128.7 (2CH), 128.5 (CH), 128.4 (C), 126.9 (CH), 125.0 (C), 123.6 (CH), 121.8 (CH), 120.2 (CH), 117.5 (C), 114.6 (CH₂), 112.2 (CH), 104.7 (2CH), 60.6 (CH₃), 56.5 (2CH₃); HRMS (ESI) $(M + H)^{+}$: *m/z* calcd for C₃₁H₂₈NO₃ 462.2069; found 462.2045.

9-(1-(2,4,6-Trimethoxyphenyl)vinyl)-9H-carbazole 4af. Flash chromatography on silica gel (EtOAc/cyclohexane 1/99) afforded 172 mg of 4af (0.48 mmol, yield 96%); white solid; mp: 134−136 °C; TLC: $R_f = 0.48$ (EtOAc/cyclohexane, 20/80, SiO₂); IR (neat) 1604, 1581, 1451, 1414, 1332, 1226, 1204; ¹H NMR (300 MHz, C₆D₆) δ (ppm) 8.06 (dd, J = 7.7, 0.7 Hz, 2H), 7.76 (d, J = 8.3 Hz, 2H), 7.39 $(ddd, J = 8.3, 7.2, 1.2 Hz, 2H), 7.21 (td, J = 7.6, 0.9 Hz, 2H), 5.92 (s,$ 2H), 5.75 (s, 1H), 5.71 (s, 1H), 3.24 (s, 3H), 3.00 (s, 6H); 13C NMR (75 MHz, C_6D_6) δ (ppm) 162.0 (C), 160.2 (2C), 141.6 (2C), 135.7 (C), 125.8 (2CH), 123.9 (2C), 120.3 (2CH), 119.7 (2CH), 117.2 (CH₂), 111.6 (2CH), 109.1 (C), 91.2 (2CH), 55.2 (2CH₃), 54.7 (CH₃); HRMS (ESI) $(M + Na)^+$: m/z calcd for C₂₃H₂₁NNaO₃ 382.1419; found 382.1396.

2-((tert-Butyldimethylsilyl)oxy)-9-(3,3-dimethylbut-1-en-2-yl)-9Hcarbazole 4ag. Flash chromatography on silica gel (EtOAc/ cyclohexane 1/99) afforded 139 mg of 4ag (0.37 mmol, yield 73%); colorless oil; TLC: $R_f = 0.69$ (EtOAc/cyclohexane, 5/95, SiO₂); IR (neat) 1625, 1599, 1495, 1457, 1345, 1284, 1253, 1224; ¹H NMR $(300 \text{ MHz}, \text{C}_6\text{D}_6)$ δ 8.98 (m, 1H), 7.88 (d, J = 8.4 Hz, 1H), 7.33 (m, 1H), 7.25 (m, 2H), 7.02 (d, J = 1.9 Hz, 1H), 6.93 (dd, J = 8.4, 2.1 Hz, 1H), 5.31 (s, 1H), 4.87 (s, 1H), 1.06 (s, 9H), 1.03 (s, 9H), 0.25 (s, 3H), 0.24 (s, 3H); ¹³C NMR (75 MHz, C_6D_6) δ 155.1 (C), 153.7 (C), 144.1 (C), 143.1 (C), 124.8 (CH), 123.8 (C), 121.1 (CH), 119.8 (CH) , 119.8 (CH), 118.2 (C), 116.7 (CH₂), 113.4 (CH), 111.0 (CH), 102.3 (CH), 39.0 (C), 30.4 (CH3), 26.0 (3CH3), 18.6 (C), −4.1 (CH_3) , −4.2 (CH₃); HRMS (ESI) (M + H)⁺: m/z calcd for $C_{24}H_{34}NOSi$ 380.2410; found 380.2409.

3,6-Dibromo-9-(1-(3,4,5-trimethoxyphenyl)vinyl)-9H-carbazole 4ah. Flash chromatography on silica gel (EtOAc/cyclohexane 2/98) afforded 191 mg of 4ah (0.37 mmol, yield 74%); white solid; mp: 134−136 °C; TLC: $R_f = 0.19$ (EtOAc/cyclohexane, 10/90, SiO₂); IR (neat) 1581, 1468, 1433, 1412, 1365, 1282, 1234; ¹H NMR (300 MHz, CD_3COCD_3) δ (ppm) 8.45 (d, J = 1.8 Hz, 2H), 7.54 (dd, J = 8.8, 1.9 Hz, 2H), 7.24 (d, J = 8.7 Hz, 2H), 6.57 (s, 2H), 6.24 (s, 1H), 5.60 (s, 1H), 3.73 (s, 3H), 3.64 (s, 6H); 13C NMR (75 MHz, CD₃COCD₃) δ (ppm) 154.7 (2C), 142.7 (C), 140.8 (2C), 140.5 (C),

132.2 (C), 130.2 (CH), 125.0 (2C), 124.3 (2CH), 113.9 (CH₂), 113.7 $(2CH)$, 113.6 $(2C)$, 104.7 $(2CH)$, 60.6 $(CH₃)$, 56.5 $(2CH₃)$; HRMS (ESI) $(M + H)^+$: m/z calcd for $C_{23}H_{20}Br_2NO_3$ 517.9792; found 517.9789.

9-(1-(4-Chlorophenyl)vinyl)-2,3,4,9-tetrahydro-1H-carbazole 4ai. Flash chromatography on silica gel (cyclohexane) afforded 123 mg of 4ai (0.4 mmol, yield 80%); colorless oil; TLC: $R_f = 0.30$ (cyclohexane, SiO_2); IR (neat) 1490, 1458, 1371, 1228; ¹H NMR (300 MHz, C_6D_6) δ (ppm) 7.62 (dd, J = 7.4, 1.1 Hz, 1H), 7.22 (m, 3H), 6.91 (m, 2H), 6.73 (m, 2H), 5.38 (s, 1H), 4.96 (s, 1H), 2.70 (m, 2H), 2.20 (m, 2H), 1.63 (m, 4H); ¹³C NMR (75 MHz, C₆D₆) δ 142.3 (C), 138.1 (C), 136.4 (C), 135.7 (C), 134.9 (C), 129.1 (2CH), 128.7 (C), 127.7 $(2CH)$, 122.0 (CH), 120.3 (CH), 118.4 (CH), 112.4 (CH₂), 111.7 (C), 110.8 (CH), 23.7 (CH₂), 23.5 (CH₂), 23.2 (CH₂), 21.5 (CH₂); HRMS (APCI) $(M + H)^+$: m/z calcd for C₂₀H₁₉ClN 308.1206; found 308.1223.

1-(1-(2,4,6-Trimethoxyphenyl)vinyl)-1H-pyrrole 4aj. Flash chromatography on silica gel (EtOAc/cyclohexane 2/98) afforded 45 mg of 4aj (0.18 mmol, yield 35%); orange oil ; TLC: $R_f = 0.57$ (EtOAc) cyclohexane, 20/80, SiO₂); IR (neat) 1648, 1605, 1582, 1413, 1334, 1226; ¹H NMR (300 MHz, CD_3COCD_3) δ 6.75 (m, 2H), 6.30 (s, 2H), 6.03 (m, 2H), 5.39 (s, 1H), 4.54 (s, 1H), 3.86 (s, 3H), 3.71 (s, 6H); ¹³C NMR (75 MHz, CD₃COCD₃) δ 163.1 (C), 160.4 (2C), 138.7 (C), 119.4 (2CH), 109.6 (2CH), 101.6 (CH₂), 91.6 (2CH), 56.2 (2CH₃), 55.7 (CH₃); HRMS (ESI) $(M + H)^{+}$: m/z calcd for $C_{15}H_{18}NO_3$ 260.1287; found 260.1280.

(Z)-1-(1,2-Bis(4-methoxyphenyl)vinyl)-1H-pyrrole 4ak. Flash chromatography on silica gel (EtOAc/cyclohexane 1/99) afforded 95 mg of 4ak (0.31 mmol, yield 62%); white solid; mp: 134−136 °C; TLC: $R_f = 0.67$ (EtOAc/cyclohexane, 20/80, SiO₂); IR (neat) 1605, 1510, 1442, 1244; ¹H NMR (300 MHz, CD_3COCD_3) δ 7.30 (m, 4H), 7.08 $(d, J = 8.8 \text{ Hz}, 2\text{H}), 7.01 (d, J = 8.5 \text{ Hz}, 2\text{H}), 6.94 (m, 2\text{H}), 6.71 (d, J)$ = 8.9 Hz, 2H), 3.84 (s, 3H), 3.74 (s, 3H); 13C NMR (75 MHz, CD₃COCD₃) δ 159.6 (C), 144.8(C), 141.0(C), 133.6(C), 132.3 (2CH), 131.6 (2CH), 131.1 (C), 129.0(2CH), 128.1 (2CH), 128.0 (CH), 115.1 (2CH), 114.3 (2CH), 55.5 (CH₃), 55.5 (CH₃); HRMS (ESI) $(M + H)^+$: *m/z* calcd for $C_{20}H_{20}NO_2$ 306.1494; found 306.1495.

1-(1-(2,4,6-Trimethoxyphenyl)vinyl)-6,7-dihydro-1H-indol-4(5H) one 4al. Flash chromatography on silica gel (MeOH/cyclohexane 2/ 98) afforded 92 mg of 4al (0.28 mmol, yield 55%); white solid; mp: 167−170 °C ; TLC: R_f = 0.75 (MeOH/DCM, 10/90, SiO₂); IR (neat) 1656, 1604, 1584, 1495, 1462, 1414, 1368, 1228; ¹H NMR (300 MHz, CDCl₃) δ 6.61 (d, J = 3.1 Hz, 1H), 6.51 (d, J = 3.1 Hz, 1H), 6.14 (s, 2H), 5.44 (s, 1H), 5.25 (s, 1H), 3.85 (s, 3H), 3.72 (s, 6H), 2.65 (t, J = 6.1 Hz, 2H), 2.45 (m, 2H), 2.07 (m, 2H); 13C NMR (75 MHz, CDCl₃) δ 195.0 (C), 162.1 (C), 159.3 (2C), 144.0 (C), 136.6 (C), 123.0 (CH), 121.1 (C), 113.6 (CH₂), 108.1 (C), 105.2 (CH), 90.9 $(2CH)$, 56.1 $(2CH_3)$, 55.5 (CH_3) , 37.9 (CH_2) , 24.4 (CH_2) , 23.0 (CH₂); HRMS (ESI) $(M + H)^{+}$: m/z calcd for C₁₉H₂₂NO₄ 328.1549; found 328.1553.

1-(1-(2,4,6-Trimethoxyphenyl)vinyl)-1H-benzo[d]imidazole 4am. Flash chromatography on silica gel (MeOH/DCM 1/99) afforded 62 mg of 4am (0.20 mmol, yield 40%); white solid; mp: 167−170 °C; TLC: $R_f = 0.35$ (MeOH/DCM, 5/95, SiO₂); IR (neat) 1739, 1608, 1586, 1454, 1228, 1205; ¹H NMR (300 MHz, CD_3COCD_3) δ 7.98 (s, 1H), 7.63 (m, 1H), 7.29 (m, 1H), 7.15 (m, 2H), 6.30 (s, 2H), 5.73 (s, 1H), 5.22 (s, 1H), 3.84 (s, 3H), 3.69 (s, 6H); 13C NMR (75 MHz, CD₃COCD₃) δ 163.4 (C), 160.4 (2C), 145.4 (C), 143.9 (CH), 135.8 (C) , 123.5 (CH), 122.5 (CH), 120.7 (CH), 112.4 (CH), 111.8 (CH₂), 107.7 (C), 91.8 (2CH), 56.3 (2CH₃), 55.8 (CH₃); HRMS (ESI) (M + H)⁺: m/z calcd for C₁₈H₁₉N₂O₃ 311.1396; found 311.1396.

1-(1-(2-Chlorophenyl)vinyl)-1H-imidazole 4an. Flash chromatography on silica gel (MeOH/DCM 2/98) afforded 44 mg of 4an (0.22 mmol, yield 43%); colorless oil ; TLC: $R_f = 0.43$ (MeOH/DCM, 5/95, $SiO₂$); IR (neat) 1649, 1487, 1432, 1375, 1321, 1245; ¹H NMR (300 MHz, CD_3COCD_3) δ 7.53 (m, 5H), 7.21 (s, 1H), 7.03 (s, 1H), 5.74 $(d, J = 1.4 \text{ Hz}, 1H), 5.13 (d, J = 1.4 \text{ Hz}, 1H);$ ¹³C NMR (75 MHz, CD₃COCD₃) δ 141.5 (C), 135.6 (C), 133.9 (C), 132.9 (CH), 132.1 (2CH), 130.9 (CH), 130.3 (CH), 128.5 (CH), 118.7 (CH), 107.2

(CH₂); HRMS (APCI) (M + H)⁺: m/z calcd for C₁₁H₁₀ClN₂ 205.0533; found 205.0533.

Synthesis of d2-Deuterated N-Tosylhydrazone 6. Synthesis of d2-Deuterated Ketone. To an Emrys Optimizer 2−5 mL pyrex reaction vessel were added 1-methoxy-4-[(4-methoxyphenyl)ethynyl] benzene (1 mmol, 238 mg) and PTSA·H₂O (0.3 mmol, 57 mg) in $CD₃OD$ (3 mL). The reaction vessel was then placed in the Emrys Optimizer and exposed to microwave irradiation according to the following specifications: 120 °C during 30 min; fixed hold time, on; sample absorption, high; prestirring, 60 s. After cooling to room temperature, H_2O (5 mL) was added and the mixture was extracted with EtOAc $(3 \times 5 \text{ mL})$. Organic layers were dried and concentrated, and the crude was purified by column chromatography on silica gel (EtOAc/cyclohexane 5/95) to afford 224 mg of d_2 -deuterated ketone (225 mg, 0.87 mmol, yield 87%); ¹H NMR (300 MHz, CDCl₃) δ (ppm) 7.99 (d, $J = 8.9$ Hz, 2H), 7.19 (d, $J = 8.6$ Hz, 2H), 6.92 (d, $J =$ 8.9 Hz, 2H), 6.86 (d, J = 8.7 Hz, 2H), 3.85 (s, 3H), 3.78 (s, 3H).

 d_2 -Deuterated N-Tosylhydrazone 6. Prepared according to the general procedure with d_2 -deuterated ketone (2 mmol, 516 mg) to give the corresponding tosylhydrazone; 481 mg (60% deuterated according to NMR integration) $(1.13 \text{ mmol, yield } 57\%)$; ¹H NMR $(300 \text{ MHz}, \text{CDCl}_3) \delta \text{ (ppm)}$ 7.68 (m, 4H), 7.53 (bs, 1H), 7.27 (m, 2H), 6.88 (m, 4H), 6.73 (d, J = 8.6 Hz, 2H), 3.92 (m, 0.8H), 3.82 (s, 3H), 3.80 (s, 3H), 2.43 (s, 3H).

4ao. Prepared according to the general procedure to give the corresponding product 144 mg (35% deuterated according to NMR integration) (0.41 mmol, yield 81%); $^1\text{H NMR}$ (300 MHz, CDCl₃) δ (ppm) 7.69 (d, $J = 7.7$ Hz, 1H), 7.13 (m, 3H), 7.07 (d, $J = 7.2$ Hz, 1H), 7.03 (d, J = 3.2 Hz, 1H), 6.98 (d, J = 8.2 Hz, 1H), 6.96 (s, 0.75H), 6.84 (d, J = 8.8 Hz, 2H), 6.67 (m, 5H), 3.81 (s, 3H), 3.72 (s, 3H).

1,2,3-Trimethoxybenzene-d (Compound 8). Isolated from the crude mixture of coupling between deuterated hydrazone, indole, and 5-iodo-1,2,3-trimethoxybenzene as oxidant; ¹H NMR (300 MHz, CD₃COCD₃) δ (ppm) 6.97 (t, J = 8.4 Hz, 0.7H), 6.64 (m, 2H), 3.80 $(s, 6H)$, 3.72 $(s, 3H)$.

Computational Methods. Calculations have been carried out with the Gaussian 09 package of programs.²⁵ Full geometry optimizations for all compounds were carried out with the use of the B3LYP²⁶ density functional level of theory an[d w](#page-10-0)ith the following basis set denoted as BS1. A 6-31G(d) basis set was employed for the first- (H) [and](#page-10-0) second-row (C, N) elements. The standard LANL2DZ small-core relativistic effective-core potential with a valence shell of double- ζ quality was used on palladium.²⁷ To get accurate energies and Gibbs free energies, the SCF convergence criterion has been systematically tightened to 10^{-8} a.u., and [the](#page-10-0) force minimizations were carried out until the rms force becomes smaller than (at least) $1 \times$ 10[−]⁵ a.u. Frequency analyses were carried out to confirm that the reported structures are minima or transition states on the B3LYP/BS1 potential energy surface, and to evaluate the thermal and entropic contributions necessary for the calculation of Gibbs free energies G. Intrinsic reaction coordinate (IRC) calculations have been performed to ascertain the identity of the transition structure under consideration. The validity of this level of calculation has been demonstrated in previous studies on Pd(II) complexes.²⁸ This is confirmed for the energetic data examined in this study by comparison with improved energies. These energies were obtained [by](#page-10-0) single-point calculations, at the B3LYP/BS1 geometries, at the B3LYP and $M06L^{29}$ density functional levels of theory, with the extended def2-TZVPP basis set and its associated ECP for Pd,³⁰ denoted as BS2, whic[h h](#page-10-0)as been retrieved from the EMSL Basis Set Library.³¹

■ ASSOCIATED CONTENT

S Supporting Information

Details for experimental conditions and copies of ${}^{1}H$ and ${}^{13}C$ NMR spectra for all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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The authors declare no competing financial interest.

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■ REFERENCES

(1) Li, H.; Boonnak, N.; Padwa, A. J. Org. Chem. 2011, 76, 9488. (2) Ogata, M.; Matsumoto, H.; Shimizu, S.; Kida, S.; Shiro, M.; Tawara, K. J. Med. Chem. 1987, 30, 1348.

(3) (a) Brustolin, F.; Castelvetro, V.; Ciardelli, F.; Ruggeri, G.; Colligiani, A. J. Polym. Sci., Part A: Polym. Chem. 2001, 39, 253. (b) Jankus, V.; Monkman, A. P. Adv. Funct. Mater. 2011, 21, 3350. (c) Chemek, M.; Ayachi, S.; Hlel, A.; Wery, J.; Lefrant, S.; Alimi, K. ́ J. Appl. Polym. Sci. 2011, 122, 2391.

(4) Fridkin, G.; Boutard, N.; Lubell, W. D. J. Org. Chem. 2009, 74, 5603.

(5) For a review, see: Dehli, J. R.; Legros, J.; Bolm, C. Chem. Commun. 2005, 973.

(6) (a) Lebedev, A. Y.; Izmer, V. V.; Kazyul'kin, D. N.; Beletskaya, I. P.; Voskoboynikov, A. Z. Org. Lett. 2002, 4, 623. (b) Venkat Reddy, C. R.; Urgaonkar, S.; Verkade, J. G. Org. Lett. 2005, 7, 4427. (c) Taillefer, M.; Ouali, A.; Renard, B.; Spindler, J.-F. Chem.-Eur. J. 2006, 12, 5301. (d) Liao, Q.; Wang, Y.; Zhang, L.; Xi, C. J. Org. Chem. 2009, 74, 6371. (e) Movassaghi, M.; Ondrus, A. E. J. Org. Chem. 2005, 70, 8638. (7) (a) Verma, A. K.; Joshi, M.; Singh, V. P. Org. Lett. 2011, 13, 1630. (b) Zhang, Y.; Donahue, J. P.; Li, C.-J. Org. Lett. 2007, 9, 627. (c) For recent reviews, see: Severin, R.; Doye, S. Chem. Soc. Rev. 2007, 36, 1407. (d) Müller, T. E.; Hultzsch, K. C.; Yus, M.; Foubelo, F.; Tada, M. Chem. Rev. 2008, 108, 3795.

(8) Beller, M.; Breindl, C.; Eichberger, M.; Hartung, C. G.; Seayad, J.; Thiel, O. R.; Tillack, A.; Trauthwein, H. Synlett 2002, 2002, 1579.

(9) (a) Imahori, T.; Hori, C.; Kondo, Y. Adv. Synth. Catal. 2004, 346, 1090. (b) Wang, L.; Huang, J.; Peng, S.; Liu, H.; Jiang, X.; Wang, J. Angew. Chem., Int. Ed. 2013, 52, 1768.

(10) For synthesis of N-vinylcarbazoles by Pd-catalyzed aza-Wacker reaction of N-H carbazoles with styrenes, see: Takeda, D.; Hirano, K.; Satoh, T.; Miura, M. Org. Lett. 2013, 15, 1242.

(11) For reviews, see: (a) Barluenga, J.; Valdés, C. Angew. Chem., Int. Ed. 2011, 50, 7486. (b) Shao, Z.; Zhang, H. Chem. Soc. Rev. 2012, 41, 560.

(12) Ding, Q.; Cao, B.; Yuan, J.; Liu, X.; Peng, Y. Org. Biomol. Chem. 2011, 9, 748.

(13) Barluenga, J.; Tomás-Gamasa, M.; Aznar, F.; Valdés, C. Angew. Chem., Int. Ed. 2010, 49, 4993. Reductive etherification product (2-(1- (tert-butoxy)ethyl)-1,3,5-trimethoxybenzene).

(14) Li, H.; Wang, L.; Zhang, Y.; Wang, J. Angew. Chem., Int. Ed. 2012, 51, 2943.

(15) Hamze, A.; Treguier, B.; Brion, J.-D.; Alami, M. Org. Biomol. Chem. 2011, 9, 6200.

(16) (a) Treguier, B.; Hamze, A.; Provot, O.; Brion, J. D.; Alami, M. Tetrahedron Lett. 2009, 50, 6549. (b) Brachet, E.; Hamze, A.; Peyrat, J. F.; Brion, J. D.; Alami, M. Org. Lett. 2010, 12, 4042. (c) Rasolofonjatovo, E.; Tréguier, B.; Provot, O.; Hamze, A.; Brion, J.-D.; Alami, M. Eur. J. Org. Chem. 2012, 1603. (d) Aziz, J.; Brachet, E.; Hamze, A.; Peyrat, J.-F.; Bernadat, G.; Morvan, E.; Bignon, J.; Wdzieczak-Bakala, J.; Desravines, D.; Dubois, J.; Tueni, M.; Yassine, A.; Brion, J.-D.; Alami, M. Org. Biomol. Chem. 2013, 11, 430. (e) Roche, M.; Hamze, A.; Provot, O.; Brion, J.-D.; Alami, M. J. Org. Chem. 2013, 78, 445. (17) Roche, M.; Hamze, A.; Brion, J.-D.; Alami, M. Org. Lett. 2013, 15, 148.

(18) Bamford, W. R.; Stevens, T. S. J. Chem. Soc. 1952, 4735. Bamford−Stevens alkene: (1,3,5-trimethoxy-2-vinylbenzene).

(19) For recent leading examples of catalytic oxidation with haloarenes, see: (a) Guram, A. S.; Bei, X.; Turner, H. W. Org. Lett. 2003, 5, 2485. (b) Berini, C.; Winkelmann, O. H.; Otten, J.; Vicic, D. A.; Navarro, O. Chem.-Eur. J. 2010, 16, 6857. (c) Maekawa, T.; Sekizawa, H.; Itami, K. Angew. Chem., Int. Ed. 2011, 50, 7022.

(20) For examples of Pd carbene species, see: (a) Barluenga, J.; Moriel, P.; Valdes, C.; Aznar, F. Angew. Chem., Int. Ed. 2007, 46, 5587. (b) Xiao, Q.; Ma, J.; Yang, Y.; Zhang, Y.; Wang, J. Org. Lett. 2009, 11, 4732.

(21) The d_2 -deuterated compound (6) was prepared from deuterated ketone according to our protocol; see: Jacubert, M.; Provot, O.; Peyrat, J.-F.; Hamze, A.; Brion, J.-D.; Alami, M. Tetrahedron 2010, 66, 3775.

(22) Computational details can be found in the Supporting Information.

(23) Perrin, D. D.; Armarego, W. L. F.; Perrin, D. R. Purification of Laboratory Chemicals, 2nd ed.; Pergamon Press: Oxford, [U.K.,](#page-9-0) [1980.](#page-9-0)

[\(24\)](#page-9-0) [Crear](#page-9-0)y, X.; Tam, W. W.; Albizati, K. F.; Stevens, R. V. Org. Synth. 1986, 64, 207.

(25) Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani, G.; Barone, V.; Mennucci, B.; Petersson, G. A.; Nakatsuji, H.; Caricato, M.; Li, X.; Hratchian, H. P.; Izmaylov, A. F.; Bloino, J.; Zheng, G.; Sonnenberg, J. L.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Vreven, T.; Montgomery, J. A., Jr.; Peralta, J. E.; Ogliaro, F.; Bearpark, M.; Heyd, J. J.; Brothers, E.; Kudin, K. N.; Staroverov, V. N.; Keith, T.; Kobayashi, R.; Normand, J.; Raghavachari, K.; Rendell, A.; Burant, J. C.; Iyengar, S. S.; Tomasi, J.; Cossi, M.; Rega, N.; Millam, J. M.; Klene, M.; Knox, J. E.; Cross, J. B.; Bakken, V.; Adamo, C.; Jaramillo, J.; Gomperts, R.; Stratmann, R. E.; Yazyev, O.; Austin, A. J.; Cammi, R.; Pomelli, C.; Ochterski, J. W.; Martin, R. L.; Morokuma, K.; Zakrzewski, V. G.; Voth, G. A.; Salvador, P.; Dannenberg, J. J.; Dapprich, S.; Daniels, A. D.; Farkas, O.; Foresman, J. B.; Ortiz, J. V.; Cioslowski, J.; Fox, D. J. Gaussian 09, revision B.01; Gaussian, Inc: Wallingford, CT, 2010.

(26) (a) Becke, A. D. J. Chem. Phys. 1993, 98, 5648. (b) Lee, C. T.; Yang, W. T.; Parr, R. G. Phys. Rev. B 1988, 37, 785.

(27) Hay, P. J.; Wadt, W. R. J. Chem. Phys. 1985, 82, 270.

(28) (a) Barluenga, J.; Florentino, L.; Aznar, F.; Valdés, C. Org. Lett. 2010, 13, 510. (b) Tan, K. V.; Dutton, J. L.; Skelton, B. W.; Wilson, D. J. D.; Barnard, P. J. Organometallics 2013, 32, 1913. (c) Roesle, P.; Dürr, C. J.; Möller, H. M.; Cavallo, L.; Caporaso, L.; Mecking, S. J. *Am*. Chem. Soc. 2012, 134, 17696.

(29) (a) Zhao, Y.; Truhlar, D. G. J. Chem. Phys. 2006, 125, 194101. (b) Zhao, Y.; Truhlar, D. G. Theor. Chem. Acc. 2008, 120, 215.

(30) (a) Weigend, F.; Ahlrichs, R. Phys. Chem 2005, 7, 3297. (b) Andrae, D.; Haeusermann, U.; Dolg, M.; Stoll, H.; Preuss, H. Theor. Chim. Acta 1990, 77, 123.

(31) (a) Feller, D. J. Comput. Chem. 1996, 17, 1571. (b) Schuchardt, K. L.; Didier, B. T.; Elsethagen, T.; Sun, L.; Gurumoorthi, V.; Chase, J.; Li, J.; Windus, T. L. J. Chem. Inf. Model. 2007, 47, 1045.